## 10.40 Thermodynamics Problem Set 8

**Fall 2003** 

Problem 10.11 Text

## **Solution:**

The second virial coefficient in molar units is given by Eq. (10-114)

$$B(T) = -2\pi N_A \int_0^\infty \left[ \exp\left[\frac{-\Phi(r)}{kT}\right] - 1 \right] r^2 dr \tag{1}$$

For a hard-sphere fluid

$$\Phi(r) = \infty$$
  $r \le \sigma$   
 $\Phi(r) = 0$   $r > \sigma$ 

Substituting into Eq. (1)

$$B^{HS} = -2\pi N_A \left[ \int_0^{\sigma} (-1) r^2 dr + \int_{\sigma}^{\infty} (0) r^2 dr \right]$$

$$B^{HS} = \frac{2}{3} \pi N_A \sigma^3 \neq f(T)$$
(2)

For a square-well fluid

$$\Phi(r) = \infty \qquad r \le \sigma$$

$$\Phi(r) = -\varepsilon \qquad \sigma < r \le R\sigma$$

$$\Phi(r) = 0 \qquad r > R\sigma$$

Substituting again into Eq. (1)

$$B^{SW}(T) = -2\pi N_A \left[ \int_0^{\sigma} (-1)r^2 dr + \int_{\sigma}^{R\sigma} (\exp(+\varepsilon/kT) - 1)r^2 dr + \int_{R\sigma}^{\sigma} (0)r^2 dr \right]$$

$$B^{SW}(T) = \frac{2\pi N_A \sigma^3}{3} \left[ 1 + (1 - \exp(\varepsilon/kT))(R^3 - 1) \right] = B^{HS} g(T)$$
(3)

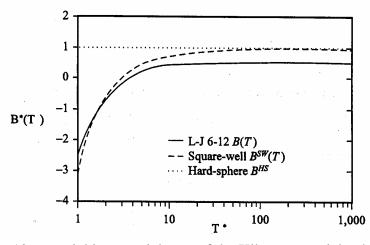
With R = 1.5

$$B^{SW}(T) = B^{HS} \left[ 1 + (1 - \exp(\varepsilon/kT))((3/2)^3 - 1) \right] = B^{HS} \left[ \frac{1}{8} (27 - 19 \exp(\varepsilon/kT)) \right]$$
(4)

Defining

$$T^* \equiv kT/\varepsilon$$
 and  $B^*(T) \equiv B(T)/B^{HS}$ 

The following figure shows the dimensionless B(T) for these potentials scaled to the temperature independent hard-sphere  $B^{HS}$ 



The Leonard Jones 6-12 potential is a special case of the Kihara potential, when  $a^* = 0$ , so we expect the Kihara potentials to behave similarly to the LJ 6-12 potential.

At 
$$T^* = 1000$$
:  $B^*_{HS} = B^*_{SW} = 1$ ,  $B^*_{L-J} = 0.29$ . At  $T^* = 1000$ :  $B^*_{HS} = B^*_{SW} = 1$ ,  $B^*_{L-J} = 0.29$ .

At  $T^*$  = infinity, the Kihara potential will go to values slightly greater than 0 depending on the value of  $a^*$ .