

## 12 Air stripping

Used to remove volatile organic compounds (VOCs), ammonia, H<sub>2</sub>S

Basic principles = mass exchange between gas and water phases

Henry's Law

$$\frac{C_g}{C_w} = H'$$

H' = dimensionless Henry's Law coeff

C<sub>g</sub> = conc in gas (moles/m<sup>3</sup>)

C<sub>w</sub> = conc in water (moles/m<sup>3</sup>)

$$\frac{P}{C_w} = H$$

P = partial pressure of gas in atm

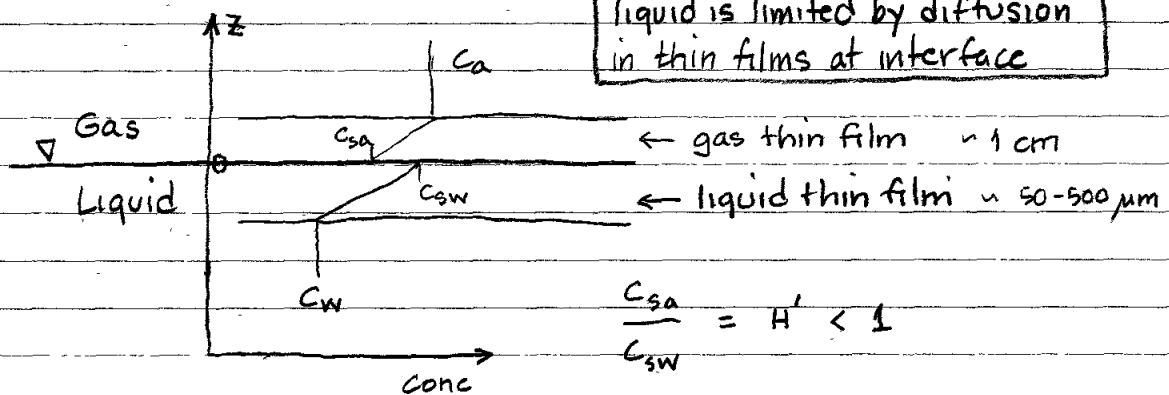
H = dimensional Henry's Law coeff  
(atm m<sup>3</sup>/mol)

$$H' = H / RT$$

R = gas const =  $8.206 \times 10^{-5}$  atm m<sup>3</sup>/mol · °K

T = absolute temp in °K

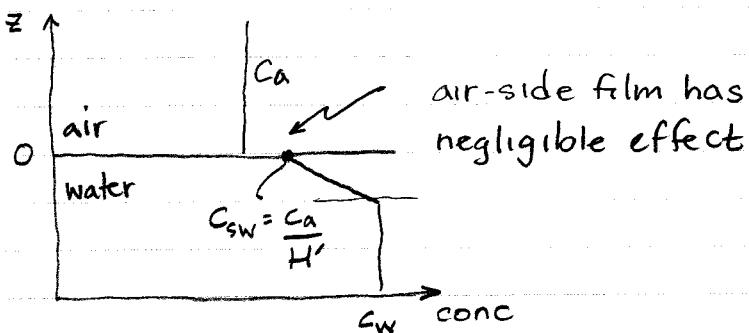
Two-film theory



For VOCs

 $H \gg 0.01$ 

only water-side film controls



$$\text{Rate of mass transfer} = \frac{dm}{dt} = - D_w A \left[ \frac{Ca/H' - Cw}{\delta_w} \right]$$

 $m$  = mass $\delta_w$  = thickness of water-side film $D_w$  = molecular diffusion coeff for water $A$  = interface area between air & waterExamples of  $H'$ 

TCE (trichloroethylene) - common industrial solvent - 0.53

Carbon tetrachloride - 0.98

 $O_2$  - 26

Benzene - 0.24

Ammonia gas - 0.73 (Note: pH must be raised to convert ionic  $NH_4^+$  (ammonium) to gaseous  $NH_3$  (ammonia):  
 $NH_3$  (percent) =  $100 / (1 - 1.75 \times 10^9 [H^+])$ )

Note: convert conc in moles/liter to conc in g/liter by multiplying by molecular weight (g/mole)

Vapor pressure defines the "saturation" concentration of chemical in a gas

V.P. = partial pressure of a chemical in a gas phase in equilibrium with the pure chemical

Example: head space in closed bottle of liquid TCE will be at V.P.

If  $VP > 1.3 \times 10^{-3}$  atm, compound is defined as volatile

Goal of treatment process design is to maximize

$$\frac{dm}{dt} = - D_w A \left[ \frac{C_a/H' - C_w}{S_w} \right]$$

$C_w$  is fixed (influent conc.)

$D_w, H'$  are essentially fixed (could change temp.)

$A$  is increased by splashing water to form smaller droplets

$S_w$  is decreased by increasing turbulence

$(C_a/H' - C_w)$  is increased by decreasing  $C_a$

Accomplished via counter-current air stripping tower - pg 4

Water with compounds to be stripping splashes down through tower film (maximizing  $A$ ), clean air is drafted upward (minimizing  $C_a$ )

# Design of an Air-Stripping Tower

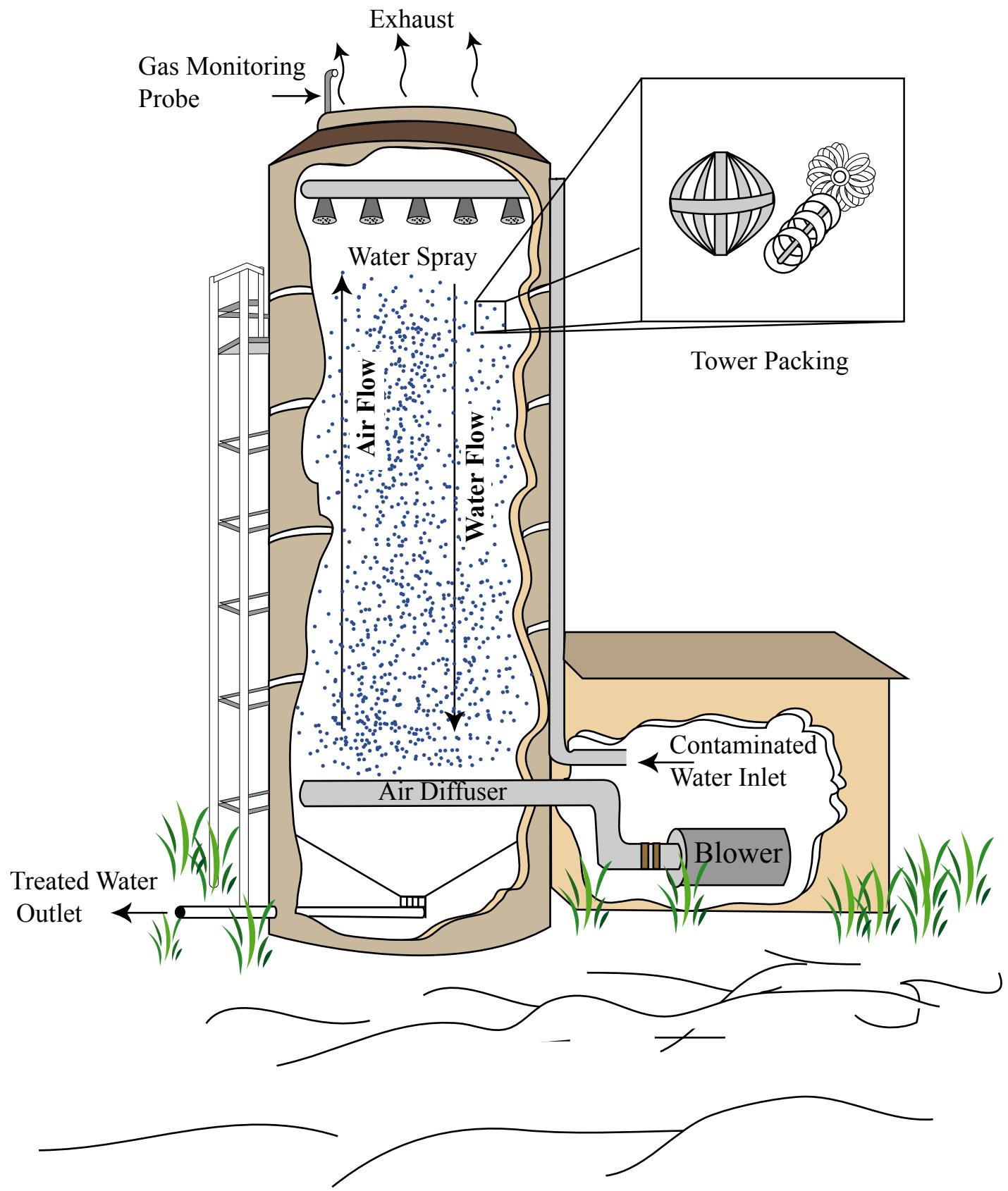
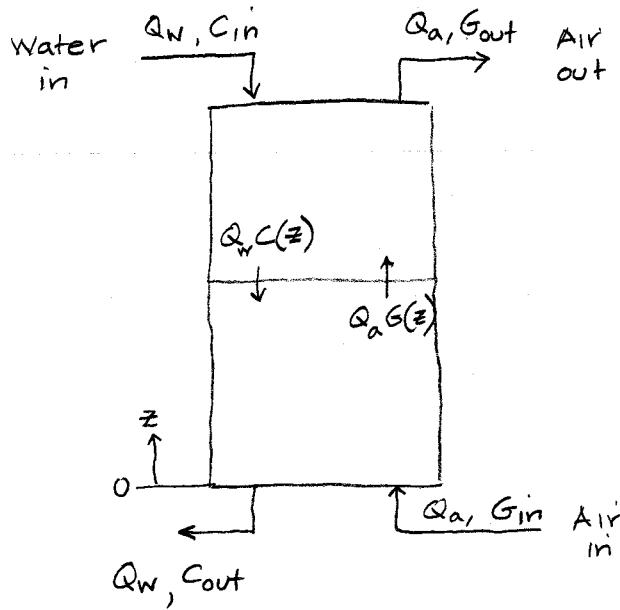


Figure by MIT OCW.

Adapted from: Fetter, C. W. *Contaminant Hydrogeology*. New York, NY: Macmillan Publishing Company, 1993, p. 418.

Mass balance for air stripper =



$Q_w$  = water flow rate  $[L^3/T]$

$Q_a$  = air flow rate  $[L^3/T]$

$C_{in}$  = influent water conc.  $[M/L^3]$

$C_{out}$  = effluent water conc.

$G_{in}$  = influent air conc  $[M/L^3]$

$G_{out}$  = effluent air conc  
(set by env'l regulations)

$z$  = height above bottom of air stripper  $[L]$

$C(z)$  = water conc at  $z$

$G(z)$  = air conc at  $z$

Mass balance between 0 and  $z$ :

$$\text{Mass in} \quad \text{Mass out} \\ Q_w C(z) + Q_a G_{in} = Q_w C_{out} + Q_a G(z) \quad (1)$$

Assume  $G_{in} = 0$

$$G(z) = \frac{Q_w}{Q_a} (C(z) - C_{out}) \quad (2)$$

For overall air stripper

$$G_{out} = (Q_w/Q_a) (C_{in} - C_{out}) \quad (3)$$

With equilibrium per Henry's Law

$$G_{out} = H' C_{in} = \frac{Q_w}{Q_a} (C_{in} - C_{out}) \quad (4)$$

Defines minimum air to water flow rate ratio

$$\left( \frac{Q_a}{Q_w} \right)_{min} = \frac{C_{in} - C_{out}}{H' C_{in}} \approx \frac{1}{H'} \quad (5)$$

$$\text{Stripping factor, } S \equiv \frac{Q_a}{Q_w} H' \quad (6)$$

= number of minimum air-to-water ratios needed for high efficiency stripping

In ideal case,  $S = 1$

Practically,  $S = 2$  to 10, 3.5 is optimal

If  $S < 1$ , air stripper cannot achieve desired removal

Required air stripper height is function of mass transfer kinetics:

Mass balance for differential element of length  $\Delta z$  at height  $z$  inside tower:

$$Q c(z + \Delta z) - Q c(z) = \frac{dm'}{dt} a \Delta t \quad (7)$$

↑                      ↑  
 contaminant        mass in  
 mass in water      water  
 inflow                outflow

$\frac{dm'}{dt}$  = mass flux per unit area across air-water interface  $\left[ \frac{M}{L^2 T} \right]$

$a$  = interface area per unit volume of tower  $[L^2/L^3]$

$\Delta t$  = volume in differential element  $[L^3]$   
 $= A_T \Delta z$

$A_T$  = cross-section area of tower  $[L^2]$

$$\begin{aligned} \text{check units: } \frac{L^3}{T} \cdot \frac{M}{L^3} - \frac{L^3}{T} \cdot \frac{M}{L^3} &= \frac{M}{L^2 T} \frac{L^2}{L^3} L^3 \\ \frac{M}{T} &= \frac{M}{T} \quad \checkmark \end{aligned}$$

From thin-film theory with water-side control:

$$\frac{dm'}{dt} = -D_w \frac{G(z)/H' - C(z)}{S_w} \quad (8)$$

$$= \frac{D_w}{S_w} (C(z) - C_{eq}(z)) \quad (9)$$

$$= K_L (C(z) - C_{eq}(z)) \quad (10)$$

$K_L$  = liquid-phase mass transfer coeff  
(piston velocity) [L/T]

$C_{eq}$  = water conc in equilibrium  
with gas conc. =  $G/H'$

Back to mass balance:

$$Q_w c(z + \Delta z) - Q_w c(z) = K_L (C(z) - C_{eq}(z)) \alpha A_t \Delta z \quad (11)$$

$$\frac{Q_w}{A_t K_L \alpha} \frac{c(z + \Delta z) - c(z)}{\Delta z} = c(z) - C_{eq}(z) \quad (12)$$

In limit as  $\Delta z \rightarrow 0$

$$\frac{Q_w}{A_t K_L \alpha} \frac{dc}{dz} = c(z) - C_{eq}(z) \quad (13)$$

$$\frac{Q_w}{A_t K_L \alpha} \frac{dc}{c(z) - C_{eq}(z)} = dz \quad (14)$$

$$\frac{Q_w}{A_t K_L \alpha} \int_{C_{out}}^{C_{in}} \frac{dc}{c - C_{eq}} = \int_0^L dz = L \quad \text{req'd tower height} \quad (15)$$

To solve, need  $C_{eq}$  as function of  $C$

From Eq (2) =

$$G(z) = \frac{Q_w}{Q_a} (C(z) - C_{out}) \quad (2)$$

$$C_{eq}(z) = \frac{G(z)}{H'} = \frac{(Q_w/Q_a)(C(z) - C_{out})}{H'} \quad (16)$$

$$\therefore L = \frac{Q_w}{A_t K_L a} \left\{ \frac{C_{in}}{C_{out}} \frac{\frac{dc}{c - (Q_w/Q_a)(C - C_{out})/H'}}{c - (Q_w/Q_a)(C - C_{out})/H'} \right\} \quad (17)$$

$$\begin{aligned} &= \frac{Q_w}{A_t K_L a} \left\{ \frac{C_{in}}{C_{out}} \frac{\frac{dc}{c [1 - (Q_w/Q_a)/H'] + C_{out}(Q_w/Q_a)/H'}}{c [1 - (Q_w/Q_a)/H'] + C_{out}(Q_w/Q_a)/H'} \right\} \\ &\text{skip in class} \\ &= \frac{Q_w}{A_t K_L a} \left[ \frac{1}{1 - (Q_w/Q_a)/H'} \right] \ln \left[ c \left( 1 - \frac{Q_w/Q_a}{H'} \right) + C_{out} \frac{Q_w}{Q_a} \frac{1}{H'} \right] \left\{ \frac{C_{in}}{C_{out}} \right\} \end{aligned}$$

$$L = \frac{Q_w}{A_t K_L a} \left[ \frac{1}{1 - (Q_w/Q_a)/H'} \right] \ln \left[ \frac{C_{in} + (C_{out} - C_{in})(Q_w/Q_a)/H'}{C_{out}} \right] \quad (18)$$

since  $S = (Q_a/Q_w) H'$  = stripping factor

$$L = \frac{Q_w}{A_t K_L a} \left( \frac{S}{S-1} \right) \ln \left[ \frac{1 + (C_{in}/C_{out})(S-1)}{S} \right] \quad (19)$$

For design, stripper tower is represented as a stack of transfer units:

$$L = HTU \cdot NTU$$

$$HTU = \text{height of transfer unit} = \frac{Q_w}{A_T K_L a}$$

$$\text{Generally } \frac{Q_w}{A_T} \leq 20 \frac{\text{gpm}}{\text{ft}^2} = 0.014 \frac{\text{m}}{\text{s}}$$

Manufacturer can supply  $K_L a$  values vs. temperature and flow rate (but best to test in pilot studies before final design)  $K_L a = 0.01 \text{ to } 0.05 \text{ sec}^{-1}$  for VOCs

Use  $\frac{Q_w}{A_T} = 20 \frac{\text{gpm}}{\text{ft}^2}$ , known  $Q_w$  to find  $A_T$

Use  $K_L a$  data,  $Q_w/A_T$  to find HTU

NTU = number of transfer units

$$= \frac{s}{s-1} \ln \left[ \frac{c_{in}}{c_{out}} \left( \frac{s-1}{s} \right) + \frac{1}{s} \right]$$

Design graph (pg. 10) gives fraction removed  $\left( \frac{c_{in} - c_{out}}{c_{in}} \right)$  vs. S and NTU

Note marginal decrease in NTU for  $S > 3$

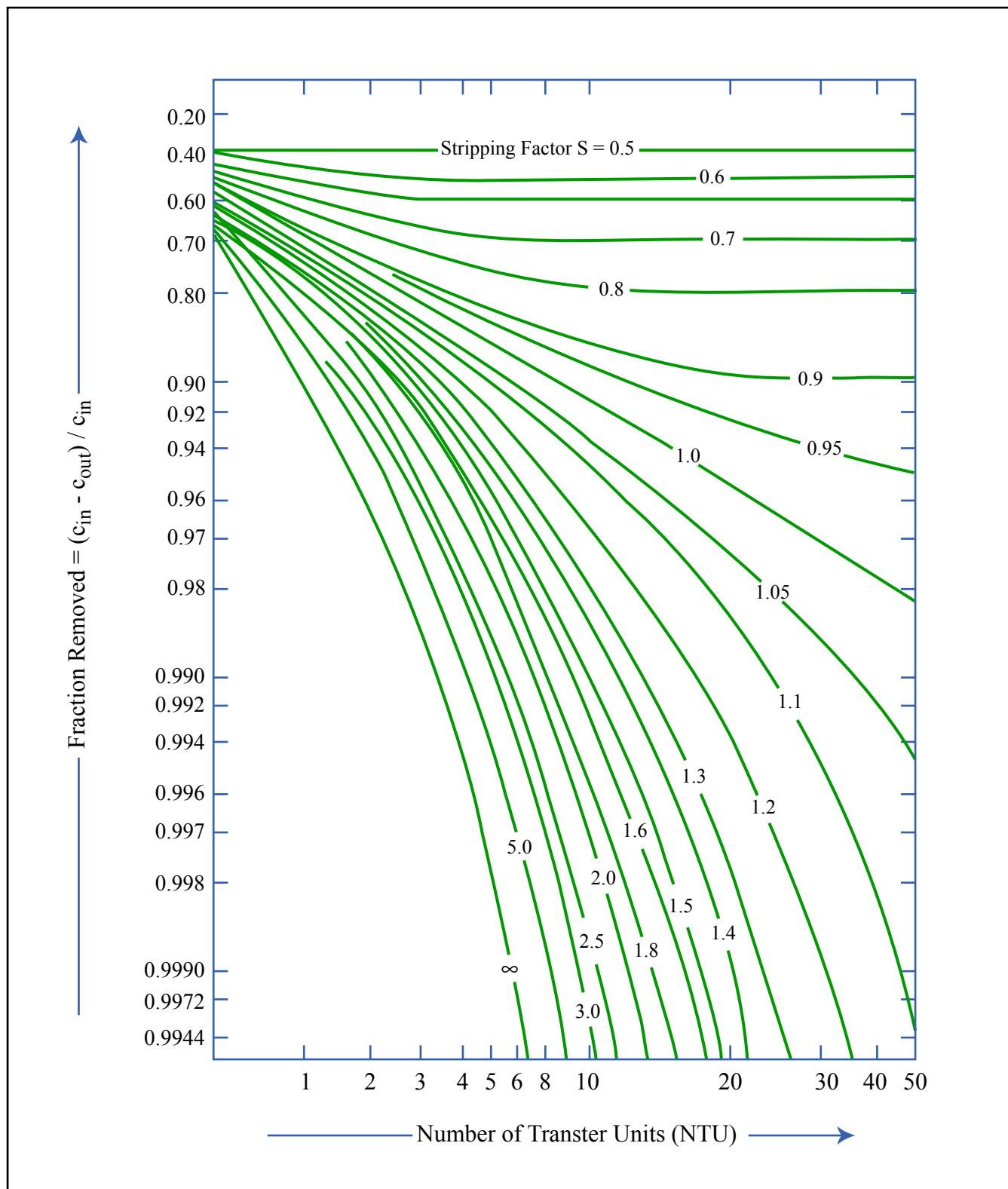


Figure by MIT OCW.

Adapted from: AWWA, 1999. Water Quality & Treatment, A Handbook of Community Water Supplies, Fifth Edition. McGraw-Hill, New York.

Full design procedure is given by:

Kavanaugh, M.C. and Trussell, R.R., 1980. Design of aeration towers to strip volatile contaminants from drinking water. Journal AWWA Vol 72 No 12 pp 684-692, December 1980.

MWH, 2005 gives same procedure

Design procedure considers pressure drop for air flow through tower, a cost factor in that blower power consumption.

Kuo, 1999 (handout) gives simplified procedure:

Get  $H'$  for contaminant of concern

Select  $s$  between 2 and 10 -  $s = 3.5$  is good estimate

$$\text{Compute } \frac{Q_a}{Q_w} = \frac{s}{H'}$$

From known  $Q_w$  find  $A$  such that  $\frac{Q_w}{A} \leq 20 \frac{\text{gpm}}{\text{ft}^2} = 0.014 \frac{\text{m}}{\text{s}}$   
 $= 14 \text{ L/sec/m}^2$

Determine desired treated water conc.,  $C_{out}$

From known  $C_{in}$ , desired  $C_{out}$  and estimated  $s$ ,  
compute

$$NTU = \left( \frac{s}{s-1} \right) \ln \left[ \frac{C_{in}}{C_{out}} \left( \frac{s-1}{s} \right) + \frac{1}{s} \right]$$

From manufacturer data for  $K_{la}$ , compute

$$HTU = \frac{Q_w}{A K_{la}}$$

Compute tower height  $L = NTU \cdot HTU$